

Longer field
and water log;
a comparison of two dissimilar
bodies of water

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The purpose of this paper is to present and compare data on the water chemistry and plankton populations of two dissimilar bodies of water. The two subjects picked, Beigner Lake and Beaver Bog, are located within the boundaries of the University of Notre Dame Environmental Research Center (UNDEC) on the Michigan/Wisconsin border about 25 miles west of Sand O'Leary, Wisconsin. Both subject waters lie within Gogebic County, Michigan proper. All information was gathered on 29 July (Beaver Bog) and 30 July (Beigner Lake), 1982, and during mid-'83 for both lakes.

The first subject, Beigner Lake, is a fairly large (~13ha) and very productive body of water. It is shallow throughout most of its area, and along its major axis, roughly north to south, it is susceptible to considerable wind-borne wave action. Beigner drains into an adjacent north area, which is called Winstone Lake, and then on into Tenmile Creek. This is all part of the Ontonagon River watershed that eventually drains into Lake Superior. Beigner is edged by thick mats of shrub growth, especially on the west side towards Winstone Lake, and surrounded by large stands of black spruce, white pine, and jack pine. Beigner is not fed by any stream but is replenished by runoff from the surrounding gentle terrain.

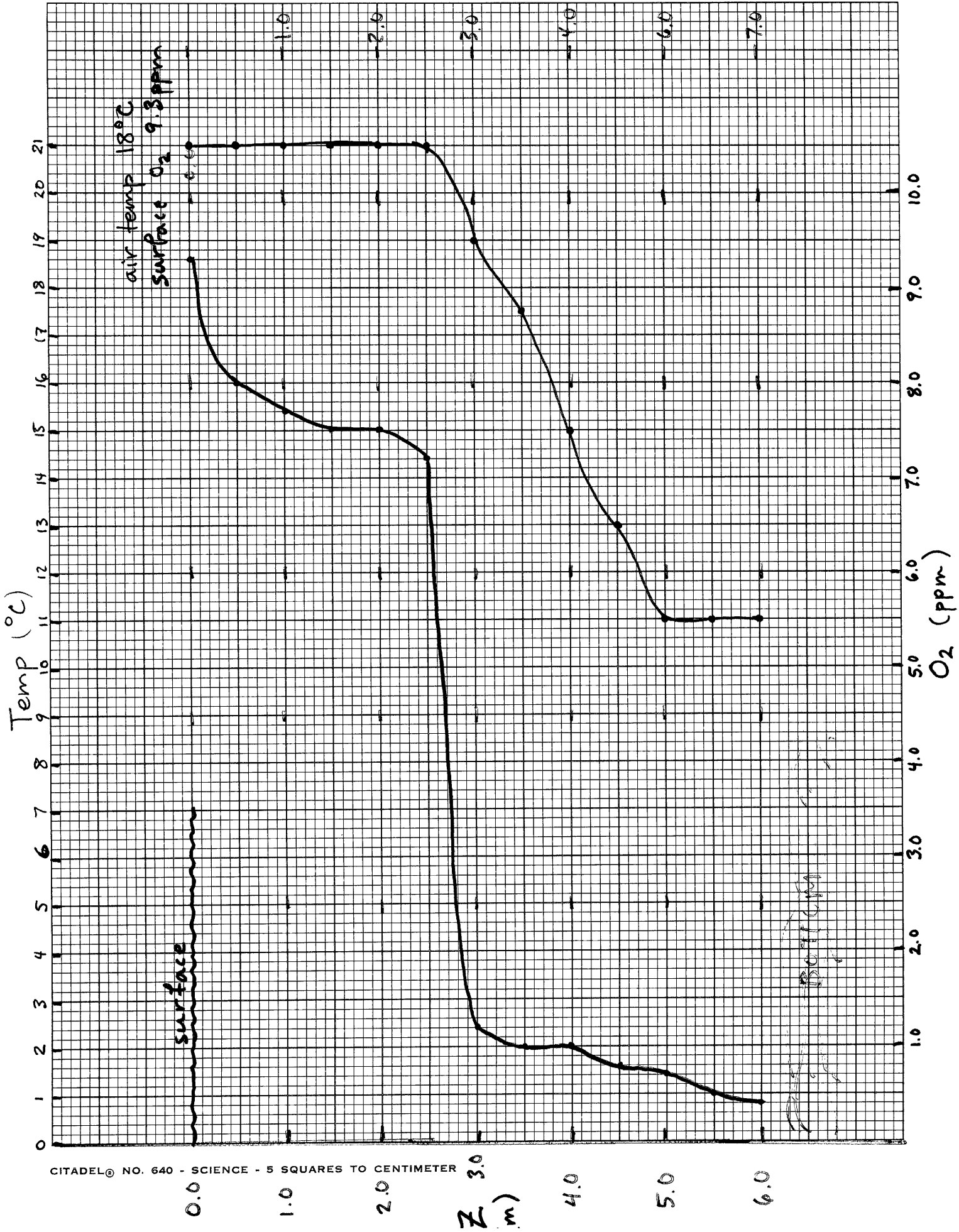
Bogner Lake

Water Chemistry

	Epilimnion	Hypolimnion
Acidity (mg/L)		
Methyl Orange	0	0
Phenolphthalein	5	15
Alkalinity (mg/L)	6	6
Color		
Apparent	86	
True	72	
Hardness (mg/L)		
Ca ⁺⁺	3	3
Mg ⁺⁺	0	0
Total	3	3
Nitrates (mg/L)	0.7	0.5
Phosphates (mg/L)		
Ortho	0.08	0.17
Total	0.20	0.30
Specific Conductance (μ mho/cm)	16	18
Sulfates (mg/L)	0.1	0.1
Sulfide	negative	negative
pH		
at lake (indicator paper)	5.0	5.5
in lab (Hach kit)	5.8	6.0
Secchi disk (m)	1.5	
Iron (mg/L)	0.08	0.14

Temperature and Oxygen curves on next page.

BERGNER LAKE O₂ and Temp. curves



Doragne Lake

Plankton

Phytoplankton: Concentration (from net)

Peridinium	1260/ml
Microcystis	900/ml
Staurastrum	240/ml
Dinobryon	300/ml
Asterionella	40/ml
Anacystis	*
Arthrodesmus	*
Desmidium	*

Zooplankton:

Keratella	120/ml
Holopedium	120/ml
Diaptomus	120/ml
Cyclops	110/ml
<u>Scaphium</u> larvae	90/ml
Tubocina	*
Kellicottia	*
Therocanus (in live sample 2 weeks later)	

* taken samples

Beaver Bog, our second subject, differs a great deal from Bogner's bog. It is a seepage bog, receiving runoff only from the immediate area and losing water only through evaporation and underground seepage. Beaver Bog is surrounded by a large Sphagnum mat which supports a characteristic bog-like vegetative cover (mostly leatherleaf, black spruce and tamarack). This dense stand of stunted conifers on the mat and the surrounding maple and aspen forest shut off Beaver Bog from almost all outside wind. The bog does become quite cold and heavily fogged in at times as cooler air sinks into this low spot from the surrounding area.

scientific names

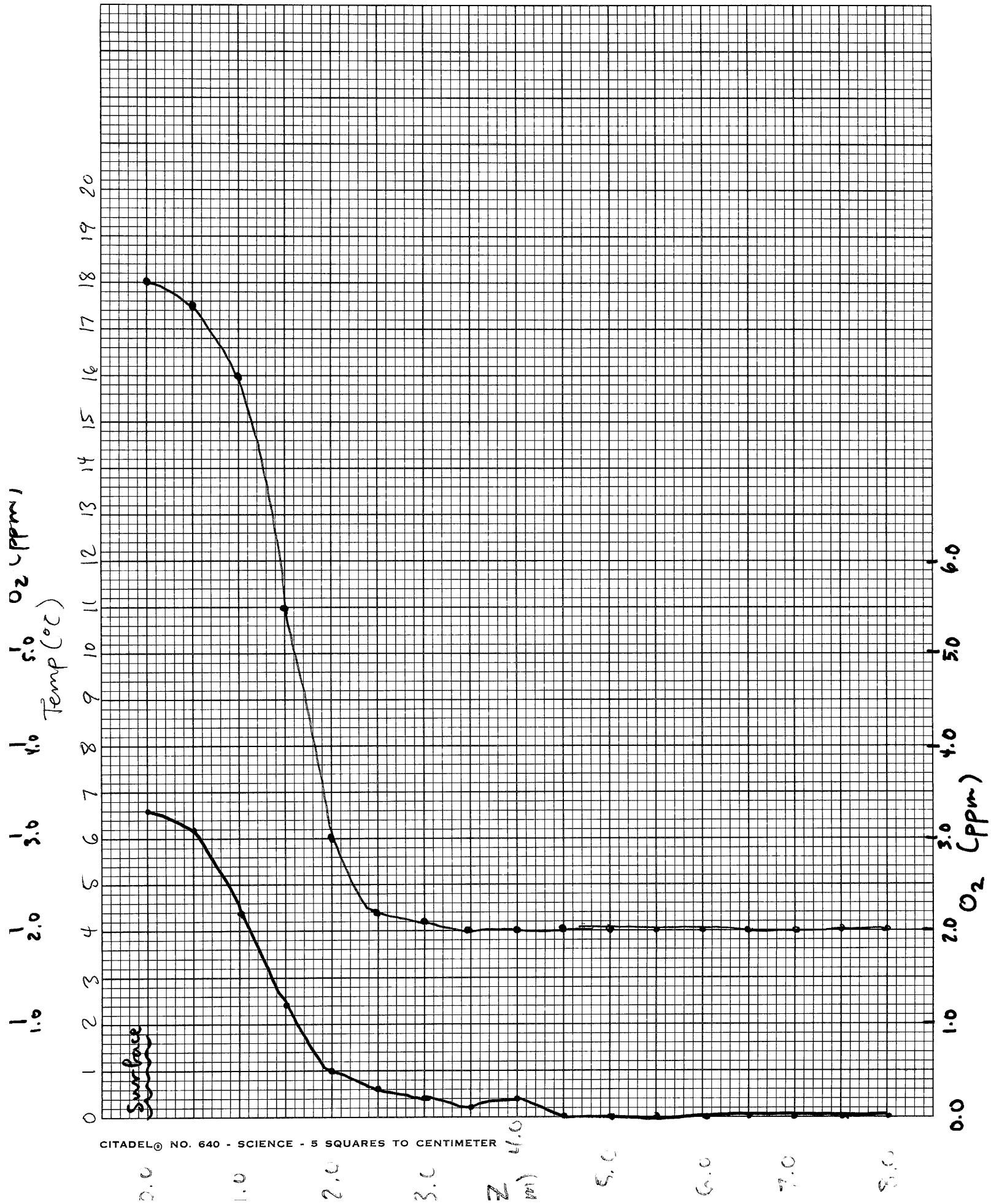
Beaver Bog

Water Chemistry

	Epilimnion	Hypolimnion
Acidity (mg/L)		
Methyl Orange	0	0
Phenolphthalein	50	70
Alkalinity (mg/L)	50	60
Color		
Apparent	210	210
True	170	160
Hardness (mg/L)		
Ca ⁺⁺	50	60
Mg ⁺⁺	10	10
Total	60	70
Nitrates (mg/L)	0.4	2.2
Phosphates (mg/L)		
Ortho	0.02	0.06
Total	0.13	0.06
Specific Conductance	21	22
Sulfates (mg/L)	0	0
Sulfide		positive
pH (in lab)	4.6	5.2
Secchi disk (m)	0.8	
Iron (mg/L)	0.22	0.36

Temperature and Oxygen curves on next page.

BEAVER BOG O₂ and Temp. curves



Draver Bog

Plankton

Phytoplankton	Concentration (from net)
Dinobryon	73500/ml
Asterionella	1160/ml
Peridinium	630/ml
Stauroctonus	210/ml
Spirogyra	*
Trigilaria	*
Zooplankton	
Daphnia	9240/ml
Polyarthra	1800/ml
Melospilum	840/ml
Tropidocyclops	740/ml
Hydrachne	300/ml
Kollicobbia	*
Water floaters	

In looking at and comparing the physical/chemical data from Bergner Lake and Beaver Bog we can notice several differences. Most prominent are the oxygen-temperature profiles of both. Bergner, a relatively shallow lake, is considerably warmer than Beaver. Its surface temperature is three degrees warmer, its thermocline is located roughly two meters deeper, and it never reaches a temperature below 11°C. In contrast to Beaver, most of Bergner's volume is contained within the epilimnion. We might say that Bergner has no true hypolimnion. Its thermocline intercepts the bottom of the lake before it ever reaches 4°C, the point of maximum water density. This formation of a deeper thermocline and general warming of the lake is obviously aided by the better mixing of the lake's larger surface as compared to the relatively calm bog.

Bergner contains considerably more dissolved oxygen than Beaver. This is again partially explained by the lake's greater wind mixing effect, but more importantly by the proliferation of photosynthetic organisms to a much greater depth than in Beaver. (Note that both Secchi disk readings were taken in less than ideal conditions: overcast skies on both, rain dimpled surface on Beaver, and wind driven waves on Bergner.) We found that Beaver was without any dissolved oxygen for much of its depth. This was confirmed by the presence of hydrogen sulfide in the hypolimnion. And interestingly enough, this explains the relatively high concentration of iron in solution. The bivalent, ferrous ion is soluble,

but only under anaerobic conditions (1,2/1).

The relative scarcity of phytoplankton nutrients and pronounced stratification of Beaver Bog lead me to believe that it did not turn over during the ~~previous~~ seasonal change. Not especially the accumulation of nitrate in the hypolimnion of Beaver. Since the concentration of this solute is largely dependent on metabolic factors, its presence is a good indication of the lack of bacterial decomposition in the depths of this bog. The lack of oxygen, combined with the extreme summer stratification and shallow penetration of light would preclude most life from the deep zone of Beaver. And in fact the lack of food and oxygen beneath the ice cover of winter prevents any fish from living there.

Beaver Bog's tea color is the result of the decomposition of allochthonous material, in particular conifer needles. The resultant organic acids are highly soluble in water, such that the centrifuged particles' true color reading was only slightly less than the apparent color which includes suspended particles.

I have covered most of the important chemical differences between these two subjects with the exception of hardness. I can find no explanation for the large amounts of dissolved calcium and magnesium ion in Beaver Bog. Both of these bodies of water rest upon or rather within pockets in the Canadian Shield. This glacier scarred granite bedrock contains very little if any limestone deposits that might

*Almbra
found in
Beaver*

contribute calcium to these lakes. Neither of them are well buffered (by bicarbonate salts) as is evident in their respectively high hydrogen~~ion~~ concentrations. In¹ furthermore, the typical physiological function of a Sphagnum mat would be to remove calcium and magnesium ions from the water for metabolic processes, replacing them with hydrogen ions (protons) to restore the water's ionic balance. I would consider these figures to be in error rather than characteristic of Beaver Bog.

I was intrigued by the diversity and abundance of plankton in both of my subject waters. Bergner, in representing a productive, soft water, oligotrophic lake, contained a great number of species and a fairly even numerical distribution. Only two genera were exceedingly numerous. These were Tarbinium, a dinoflagellate, and Microcystis, a blue green. Bergner's microfauna was primarily copepods, which thrive on algae and in turn feed small fish, insect larvae and other aquatic invertebrates.

Beaver Bog provided quite a contrast, containing a virtual profusion of two genera. Dinobryon, a flagellate, completely engulfed the counting slide and necessitated a dilution of the sample before I could even begin to estimate its number. A cladoceran, Boeckia, also accounted for a large portion of the sample, suggesting perhaps that a bloom in phytoplankton can be accompanied by a bloom in the primary consumers of that phytoplankton. But whether this is a temporal

uprange in the flux of a predator-prey population curve, or
a semi-stable relationship, that is characteristic of Proulx
(1968), I cannot say.

Reference:

Ward, George H. and Michael D. Wood,
The Ecology of Inland Waters and Estuaries.
J. Van Nostrand Co. New York, 1976.